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Equation of state of a synthetic ulvöspinel, \((\text{Fe}_{1.94}\text{Ti}_{0.03})\text{Ti}_{1.00}\text{O}_{4.00})\), at ambient temperature

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Abstract Using a diamond-anvil cell and synchrotron X-ray diffraction, the compressional behavior of a synthetic ulvöspinel \((\text{Fe}_{1.94}\text{Ti}_{0.03})\text{Ti}_{1.00}\text{O}_{4.00})\) has been investigated up to about 7.05 GPa at 300 K. The pressure–volume data fitted to the second-order Birch–Murnaghan equation of state yield an isothermal bulk modulus \((K_T)\) of 147(4) GPa \((K'_T\) fixed as 4). This value is slightly larger than that previously determined by an ultrasonic pulse echo method \((121(2)\) GPa; Syono et al., J Phys Soc Jpn 31:471–476, 1971), but substantially smaller than that recently determined by a synchrotron X-ray diffraction technique \((251(3)\) GPa; Yamanaka et al., Phys Rev B 80:134120, 2009; Am Mineral 98:736–744, 2013). Combined with the \(K_T\) of magnetite \((\text{Fe}_3\text{O}_4; \sim 182(3)\) GPa), our finding suggests that the bulk modulus of the solid solutions \(\text{Fe}_{3-x}\text{Ti}_x\text{O}_4\) \((0 \leq x \leq 1)\) along the join magnetite–ulvöspinel decreases by ~20 %.

Keywords Compressibility · Diamond-anvil cell · Synchrotron X-ray diffraction · Ulvöspinel

Introduction

Natural ulvöspinel is a common accessory mineral in terrestrial metamorphic and igneous rocks, and is an abundant phase in the high Ti environment of the lunar crust (Cameron 1971). With small amounts of additional \(\text{Mg}_2\text{TiO}_4\) and \(\text{Mn}_2\text{TiO}_4\), its compositions mainly fall along the join magnetite \((\text{Fe}_3\text{O}_4)–\text{ulvöspinel}\) \((\text{Fe}_2\text{TiO}_4)\). The solid solutions along this join, known as titanomagnetites and with the spinel structure, are important minerals not only because of their applications as geothermometer and oxygen fugacity indicator (Buddington and Lindsley 1964; Spencer and Lindsley 1981; Andersen and Lindsley 1988), but also because of their contributions in the magnetic signals of many rocks such as the mid-ocean ridge basalts. Therefore, many experimental studies were conducted to investigate the physical and chemical properties of the titanomagnetites (e.g., Ishikawa et al. 1971; O’Neill et al. 1988; Bosi et al. 2008; Kyono et al. 2011; Lilova et al. 2012; Diego Gatta et al. 2014).

Compared to the numerous experimental investigations on the bulk modulus of magnetite (e.g., Haavik et al. 2000; Reichmann and Jacobsen 2004; Diego Gatta et al. 2007; Rozenberg et al. 2007; Yamanaka et al. 2013), the number of studies on the bulk modulus of ulvöspinel has been very limited. Further, the results obtained in these studies are currently in sharp discrepancy. Syono et al. (1971) pioneered in measuring the bulk modulus of ulvöspinel. By conducting ultrasonic pulse echo measurements on a single crystal of ulvöspinel, they obtained an adiabatic bulk modulus \((K_S)\) as 121 GPa, with an error of about 2 GPa.
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(Liebermann et al. 1977). Recently, Yamanaka et al. (2009, 2013) carried out compression experiments up to ~9 GPa on both ulvöspinel single crystal (pressure medium: ethanol–methanol–water mixture; P measurement method: ruby fluorescence method; sample size: 60 × 60 × 20 μm²) and ulvöspinel powder (pressure medium: neon; P measurement method: ruby fluorescence method; gasket material: rhenium) with a diamond-anvil cell (DAC) and synchrotron X-ray diffraction. Their obtained value for the isothermal bulk modulus of ulvöspinel was ~251(3) GPa (the first pressure derivative of the isothermal bulk modulus, , fixed as 4). In principle, and are related by the equation

where is the volumetric thermal expansion coefficient and is the Grüneisen parameter. Since the value of the is in the range of 1–3 × 10⁻⁵ K⁻¹ (Fei 1995; Wang et al. 2012) and that of is commonly less than 2 (e.g., Chopelas and Hofmeister 1991; Hofmeister and Mao 2001; Wang et al. 2002, 2003; Yong et al. 2012), the difference between the and at ambient condition, with larger than , is expected to be less than 10 GPa.

In this paper, we have investigated the equation of state of a synthetic ulvöspinel at ambient temperature using a diamond-anvil cell, coupled with synchrotron X-ray radiation. Subsequently, the obtained equation of state of ulvöspinel is compared to that of magnetite, to illustrate the variation pattern of the bulk modulus along the join Fe₃O₄–Fe₂TiO₄.

**Experimental method**

The ulvöspinel was synthesized in an open Pt capsule at 1,573 K for 24 h with a controlled oxygen fugacity of fO₂ = 10⁻¹¹. The product was once examined with electron microprobe analysis (EMP: JEOL JXA-8200) and X-ray diffraction, and its high-pressure phase transition was investigated using X-ray diffraction and Mössbauer spectroscopy up to about 24 GPa (Wu et al. 2012). In this study, we further characterized this material using scanning electron microscope (SEM; FEI Quanta 650 FEG), EMPA (JEOL JXA-8230) and synchrotron X-ray diffraction.

The high-pressure angle dispersive X-ray diffraction experiments with a symmetrical DAC were conducted at the beamline X17C, National Synchrotron Light Source, Brookhaven National Laboratory. The experimental techniques were generally identical to those used in our previous studies (Liu et al. 2008, 2009; Fleet et al. 2010). Briefly, a T301 stainless steel plate was used as the gasket, a 4:1 methanol–ethanol mixture was used as the pressure medium, and a couple of ruby chips were used as the pressure marker (the ruby fluorescence method; Mao et al. 1978). The incident synchrotron radiation beam was monochromatized to a wavelength of 0.4112 Å, and its beam size was collimated to ~25 × 20 μm². Each X-ray diffraction pattern at certain pressure was collected for about 10 min using an online CCD detector, and subsequently integrated to derive the conventional one-dimension profile using the Fit2D program (Hammersley 1996).

With a full profile refinement of the XRD data using the MDI’s program Jade 5.0 (Material Data, Inc.), the positions of the diffraction peaks 111, 220, 311, 222, 400, 440, 511 and 440 were determined and later used to derive the unit-cell parameters of the ulvöspinel at different pressures. These data-processing techniques were used in our previous investigations (He et al. 2012; Chang et al. 2013).

**Results and discussion**

Electron back-scatter images obtained with the SEM show that the synthetic product from our high-temperature experiment at 1 atm (grain size ranging from 20 to 200 μm), used as the starting material in our high-pressure compression experiments, consists of one crystalline phase only (plus a trace amount of a melt-like compound). The XRD data identified this crystalline phase as ulvöspinel, but demonstrated another crystalline phase, a very small fraction of impurity ilmenite, at the same time. We took the advantage of this ilmenite impurity and used it as an internal gauge for examining the accuracy of our result about the ulvöspinel, since previous studies gave out very similar values for the isothermal bulk modulus of ilmenite (Wechsler and Pre-witt 1984; Tronche et al. 2010).

The EMPA data (10 analyses) suggest a chemical formula of Fe₁.940(8)Ti₁.030(4)O₄.₀₀₀ (all iron assumed as Fe²⁺) for the ulvöspinel. This phenomenon, with Ti in slight excess of 1 atom per formula unit, is in contrast to some previous observations made by Ôno et al. (1968; Fe₂.₀₅Ti₀.₉₅O₄.₀₀₀) and Sedler et al. (1994; Fe₂.₀₆₈Ti₀.₉₃₂O₄.₀₀₀). Without any site occupation information, we tentatively write the ulvöspinel formula as (Fe₁.₉₄Ti₀.₀₅)Ti₁.₀₀O₄.₀₀₀, which leads to the cation substitution mechanism 2Fe²⁺ → Ti⁴⁺. It appears that a trace amount of Ti (cation vacancy as well) on the tetrahedral sites of the ulvöspinel is inevitable. The presence of some Ti on the tetrahedral sites is currently a controversial issue; Wechsler et al. (1984) and Yamanaka et al. (2013) argued that Ti only occurs on the octahedral sites, whereas Forster and Hall (1965) and Sedler et al. (1994) demonstrated small amounts of Ti (up to 18 %) occupying the tetrahedral sites. We assume in this study that the effect of this trace amount of Ti (cation vacancy as well) on the tetrahedral sites on the physical properties of the ulvöspinel is negligible.

According to Akimoto and Syono (1967), Fe₃TiO₄ ulvöspinel breaks down to FeO + FeTiO₃ at about 4 GPa and

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1,473 K, with a positive Clapeyron slope for the \( P-T \) locus of the decomposition reaction. At room temperature, it transforms into a tetragonal phase at \( \sim 7.12-8.76 \) GPa (Yamanaka et al. 2009; Wu et al. 2012; Yamanaka et al. 2013). It is thus clear that a completed \( P-T \) diagram for ulvöspinel is still not available. To avoid the cubic-to-tetragonal phase transition, consequently, we conducted our high-pressure experiments up to \( 7.05 \) GPa only (Table 1). In total, nine X-ray diffraction patterns were collected during compression, while four were collected during decompression. Typical X-ray diffraction patterns are shown in Fig. 1. No peak broadening can be observed, which verifies the efficiency of our pressure medium to maintain a hydrostatic pressure condition in our experiments (Klotz et al. 2009). In addition, the X-ray diffraction pattern obtained at the maximum experimental pressure of \( 7.05 \) GPa shows no peak-splitting for the 220 peak, suggesting that the polymorphic phase transition was not reached (Yamanaka et al. 2009; Wu et al. 2012).

Duplicated analyses using the synchrotron radiation at ambient pressure give out the unit-cell parameters of the synthetic ulvöspinel as \( a_0 = 8.5274(14) \) Å and \( V_0 = 620.09(18) \) Å\(^3\), and \( a_0 = 8.5231(19) \) Å and \( V_0 = 619.14(24) \) Å\(^3\), respectively. These values are smaller than most early measurements (from \( a_0 = 8.50(1) \) Å to \( a_0 = 8.538(3) \) Å, as summarized in Sedler et al. (1994)), but in good agreement with the most recent observations (Bosi et al. 2009; Yamanaka et al. 2013). Different compositions caused by different synthesizing conditions such as varying temperature, oxygen fugacity, and employment of a flux or not are presumably the primary sources for the fluctuation of the unit-cell parameters at ambient \( P-T \) condition. In addition, the obtained room-\( P \) unit-cell parameters for the ilmenite impurity are: \( a_0 = 5.087(5) \) Å, \( c_0 = 14.055(17) \) Å and \( V_0 = 315.03(77) \) Å\(^3\).

The effect of pressure on the unit-cell volume of the ulvöspinel is summarized in Table 1 and graphically shown in Fig. 2. A small non-linear dependence on pressure over the investigated pressure range has been observed for the unit-cell volume. Further, the volume data determined from the compression experiments and decompression experiments are well mixed, suggesting that the elastic behavior of ulvöspinel is fully reversible after compression to about 7 GPa. Compared to the data from Yamanaka et al. (2009;
2013), Fig. 2a shows that the high-pressure volumes determined in this study are always smaller, with the difference between this study and Yamanaka et al. (2009; 2013) positively correlated with pressure. As to the pressure effect on the \( V/V_0 \) ratio (Fig. 2b), in contrast, most high-pressure data but the one at 8.76 GPa from Yamanaka et al. (2009) are in excellent agreement with our data, whereas all high-pressure data from Yamanaka et al. (2013) are not.

The effect of pressure on the unit-cell parameters of the ilmenite impurity is summarized in Table 1 and graphically shown in Fig. 3. Although the data are not as accurate as those for the ulvöspinel due to the low intensity of the X-ray diffraction peaks caused by the low-volume fraction of the ilmenite, concrete correlations between the unit-cell parameters and pressure have been well demonstrated, and good agreement between this study and Wechsler and Prewitt (1984) has been obtained.

To determine the elastic parameters, the \( P-V \) data have been fitted to the second-order Birch–Murnaghan equation of state (Birch 1947) by a least-squares method:

\[
P = 3K_Tf_E(1 + 2f_E)^{\frac{3}{2}}
\]

where \( P \) is the pressure, \( K_T \) the isothermal bulk modulus, and \( f_E \) the Eulerian definition of finite strain, which is

\[
[(V/V_0)^{3/2} - 1]/2
\]

respectively. In the Eulerian definition of finite strain, \( V_0 \) is the volume at zero pressure, whereas \( V \) is the volume at high pressure. A fit with the third-order Birch–Murnaghan equation of state with variable \( K_T' \) has been tested, but generated unreasonable result due to the narrow pressure range of our experiments.

With \( K_T' \) fixed as 4, we have obtained (1) \( K_T = 148(4) \) GPa and \( V_0 = 618.91(36) \) Å\(^3\) with the ambient and compression data, (2) \( K_T = 130(10) \) GPa and \( V_0 = 619.50(70) \) Å\(^3\) with the ambient and decompression data, and (3)
$K_T = 148(4)$ GPa and $V_0 = 618.78(37)$ Å$^3$ with all data. We believe that $K_T = 148(4)$ GPa should be a good approximation to the real value of the isothermal bulk modulus of ulvöspinel. By fixing $K_T'$ to different values and calculating $K_T$, a correlation analysis of $K_T$ and $K_T'$ has also been carried out, and the result is shown in Fig. 4. The value of the $K_T$ decreases from 157(5) to 132(4) GPa as the $K_T'$ value increases from 1 to 10.

Similarly, we have fitted all $P$–$V$ data of the ilmenite impurity (Table 1), and obtained $K_T = 184(9)$ GPa and $V_0 = 314.66(22)$ Å$^3$. This value for the $K_T$ is in good agreement with Wechsler and Prewitt (1984; 177(3) GPa with $K_T'$ assumed to be 4(1)), and Tronche et al. (2010; 181(7) with $K_T' = 3(1)$), suggesting that the data in this study have high quality. In addition, a linearized second-order Birch–Murnaghan equation of state (Angel 2000) was used to obtain the parameters of the equations of state for the crystallographic axes, yielding: $d_0 = 5.083(2)$ Å and $K_{T-a} = 219(19)$ GPa for the $a$-axis, and $c_0 = 14.076(5)$ Å and $K_{T-c} = 122(7)$ GPa for the $c$-axis. Therefore, ilmenite has a relatively strong elastic anisotropy, in good agreement with Wechsler and Prewitt (1984) as well.

The $K_S$ of ulvöspinel was once determined as 121(2) GPa using an ultrasonic pulse echo method (Syono et al. 1971; Liebermann et al. 1977). Considering the accuracy of their experimental techniques in the 1970s, the result from Syono et al. (1971) is in good agreement with ours. If we assume for ulvöspinel at 1 atm and 298 K $\alpha = 2.06 \times 10^{-5}$ K$^{-1}$ (equal to that of magnetite; Fei 1995; Wang et al. 2012) and $\gamma = 2$ (a potentially maximum value), and adopt $K_T = 148$ GPa (this study), anyhow, the $K_S$ value of ulvöspinel is predicted to be ~150 GPa, according to the equation $K_S = K_T (1 + \alpha \gamma T)$. It follows that a further experimental study on the $K_S$ of ulvöspinel is desirable.

The $K_T$ of ulvöspinel was recently constrained as 251(3) using a DAC + synchrotron X-ray radiation technique (Yamanaka et al. 2009, 2013). As shown in Fig. 2b, most experimental data from Yamanaka et al. (2009) are generally compatible with our data. The experiment at 8.76 GPa, displaying no sign of the cubic-to-tetragonal phase transition, was highly possibly subject to large pressure overestimation, as suggested by the experiment at 8.56 GPa in Yamanaka et al. (2013), which already demonstrated the phase transition. With the removal of the data at 8.76 GPa, the rest data from Yamanaka et al. (2009) constrain the $K_T$.

**Table 2** Isothermal bulk modulus of spinel solid solutions Fe$_{3-x}$Ti$_x$O$_4$ at ambient $P$ and $T$

<table>
<thead>
<tr>
<th>$x$</th>
<th>$K_0$</th>
<th>$K_T'$</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>217(2)$^a$</td>
<td>4$^b$</td>
<td>Haavik et al. (2000)</td>
</tr>
<tr>
<td>0</td>
<td>186(3)</td>
<td>5.1(1)</td>
<td>Reichmann and Jacobsen (2004)</td>
</tr>
<tr>
<td>0</td>
<td>182(4)</td>
<td>3.6(8)</td>
<td>Diego Gatta et al. (2007)</td>
</tr>
<tr>
<td>0</td>
<td>180(1)</td>
<td>4$^b$</td>
<td>Diego Gatta et al. (2007)</td>
</tr>
<tr>
<td>0</td>
<td>181(1)</td>
<td>4.3(1)</td>
<td>Rozenberg et al. (2007)</td>
</tr>
<tr>
<td>0</td>
<td>220(5)</td>
<td>4$^b$</td>
<td>Yamanaka et al. (2013)</td>
</tr>
<tr>
<td>0.102</td>
<td>263(4)</td>
<td>4$^b$</td>
<td>Yamanaka et al. (2013)</td>
</tr>
<tr>
<td>0.231</td>
<td>269(5)</td>
<td>4$^b$</td>
<td>Yamanaka et al. (2013)</td>
</tr>
<tr>
<td>0.551</td>
<td>233(1)</td>
<td>4$^b$</td>
<td>Yamanaka et al. (2013)</td>
</tr>
<tr>
<td>0.624</td>
<td>275(6)</td>
<td>4$^b$</td>
<td>Yamanaka et al. (2013)</td>
</tr>
<tr>
<td>0.734</td>
<td>246(2)</td>
<td>4$^b$</td>
<td>Yamanaka et al. (2013)</td>
</tr>
<tr>
<td>0.831</td>
<td>242(2)</td>
<td>4$^b$</td>
<td>Yamanaka et al. (2013)</td>
</tr>
<tr>
<td>1</td>
<td>251(3)</td>
<td>4$^b$</td>
<td>Yamanaka et al. (2009, 2013)</td>
</tr>
<tr>
<td>1</td>
<td>121(2)$^c$</td>
<td>–</td>
<td>Syono et al. (1971)</td>
</tr>
<tr>
<td>1</td>
<td>148(4)</td>
<td>4$^b$</td>
<td>This study</td>
</tr>
</tbody>
</table>

$^a$ Numbers in parentheses representing one standard deviation

$^b$ Fixed value

$^c$ Adiabatic bulk modulus
of ulvöspinel as 152(8) GPa (uncertainty in the pressure measurements assumed as 0.01 GPa), a value in excellent agreement with our result. As to the much larger $K_T$ estimated by Yamanaka et al. (2013), it is presently unclear what the real cause was. It is noted though that the ulvöspinel sample used by Yamanaka et al. (2013) is stoichiometric, whereas ours is (Fe$_{1.94}$Ti$_{0.03}$)Ti$_{1.00}$O$_{4.00}$, with a trace amount of Ti and cation vacancy on the tetrahedral sites.

Recent experimental results about the isostructural bulk moduli of the solid solutions along the join Fe$_3$O$_4$–Fe$_2$TiO$_4$ have been listed in Table 2 and graphically shown in Fig. 5. As summarized by Haavik et al. (2000), nine earlier independent studies constrained the $K_T$ of magnetite, varying from 155 to 215 GPa. The recent experimental investigations (Haavik et al. 2000; Reichmann and Jacobsen 2004; Diego Gatta et al. 2007; Rozenberg et al. 2007; Yamanaka et al. 2013) have narrowed the variation range down to about 180–217 GPa. Haavik et al. (2000) obtained a slightly larger $K_T$ for magnetite. They used N$_2$ as the pressure-transmitting medium and carried out high-pressure experiments up to about 30 GPa, so that their result might be affected by a nonhydrostatic experimental environment, as suggested by Klotz et al. (2009). Yamanaka et al. (2013) also obtained a slightly larger $K_T$ for magnetite; their value was likely inaccurate, by the same token outlined in the preceding paragraph (the ulvöspinel case). Consequently, it seems appropriate to adopt a value of about 182(3) GPa as the $K_T$ for magnetite (Reichmann and Jacobsen 2004; Diego Gatta et al. 2007; Rozenberg et al. 2007).

Figure 5 shows that along the join Fe$_3$O$_4$–Fe$_2$TiO$_4$, the $K_T$ of the solid solutions Fe$_{3-x}$Ti$_x$O$_4$ (0 ≤ x ≤ 1) decreases from about 182(3) to 148(4) GPa, as x increases from 0 to 1. Accompanying this variation is an increase in the unit-cell volume, from about 591.8(3) Å$^3$ (Reichmann and Jacobsen 2004; Diego Gatta et al. 2007; Rozenberg et al. 2007) to 618.78(37) Å$^3$ (this study). The relative decrease in the bulk moduli along the join Fe$_3$O$_4$–Fe$_2$TiO$_4$ is about 20 %, whereas the relative increase in the unit-cell volumes is about 4 % only.

It has been well established that the $K_T$ is nearly inversely proportional to the specific volume for alkali halides, fluorides, selenides, sulfides, apatites, and 2–3 spinels (Anderson and Nafe 1965; Anderson and Anderson 1970; He et al. 2013). Our result obtained for ulvöspinel, combined with the result for magnetite, seems compatible with this relationship. On the contrary, the result from Yamanaka et al. (2013) does not fit (Fig. 5).

The exact contributor to the bulk modulus variations as shown in Fig. 5 is presently unclear. Both magnetite and ulvöspinel are generally inverse spinels (Fleet 1981; Wechsler et al. 1984), with their structural formula as $T$($Fe^{3+}$)$^{3+}$M($Fe^{2+}$+$Fe^{3+}$)O$_4$ and $T$($Fe^{2+}$)$^{2+}$M($Fe^{2+}$+$Ti^{4+}$)O$_4$, respectively (T, tetrahedral sites; M, octahedral sites). The solid solutions along the join Fe$_3$O$_4$–Fe$_2$TiO$_4$ clearly involve not only a cation substitution of Ti$^{4+}$ for Fe$^{3+}$ on the M sites, but also an electron exchange reaction between the T and M sites. In addition, this complexity is exacerbated by the well-known order–disorder phenomenon in the spinel structure due to variation of temperature (Liu and Prewitt 1990; O’Neill et al. 2003), pressure (Antao et al. 2005; Rozenberg et al. 2007) or composition (Wechsler et al. 1984; Diego Gatta et al. 2014). Further, the stoichiometry of the solid solutions along the join Fe$_3$O$_4$–Fe$_2$TiO$_4$ is highly sensitive to the oxygen fugacity in their synthesizing experiments (Taylor 1964). Unfortunately, both the exact synthesizing temperature and oxygen fugacity and the detailed chemical compositions for the solid solutions used in Yamanaka et al. (2009 and 2013) are not available. More investigation on the solid solutions between magnetite and ulvöspinel is obviously necessary if one aims to understand the cause to the correlations between the $K_T$ and chemical composition, as shown in Fig. 5.

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